

$(\eta^6\text{-Arene})$ tricarbonylchromium and $(\eta^5\text{-cyclohexadienyl})$ tricarbonylmanganese complexes: indirect nucleophilic substitutions

Françoise Rose-Munch*, Vanessa Gagliardini, Christophe Renard,
Eric Rose¹

*Laboratoire de Synthèse Organique et Organométallique, UMR CNRS 761 1, Case 181,
Université P. et M. Curie, 4 place Jussieu, Tour 44, 75252 Paris, Cedex 05, France*

Received 15 December 1997; accepted 27 January 1998

Contents

Abstract	249
1. Introduction	250
2. Arenetricarbonylchromium complexes	250
2.1. <i>Tele-meta</i> nucleophilic aromatic substitution	251
2.2. <i>Cine</i> nucleophilic aromatic substitution	257
2.3. <i>Tele-para</i> nucleophilic aromatic substitution	258
3. Cationic arenetricarbonylmanganese complexes	260
3.1. Reactivity of cationic arenetricarbonylmanganese complexes	260
3.2. Reactivity of neutral $(\eta^5\text{-cyclohexadienyl})$ tricarbonylmanganese complexes	263
Acknowledgements	265
References	266

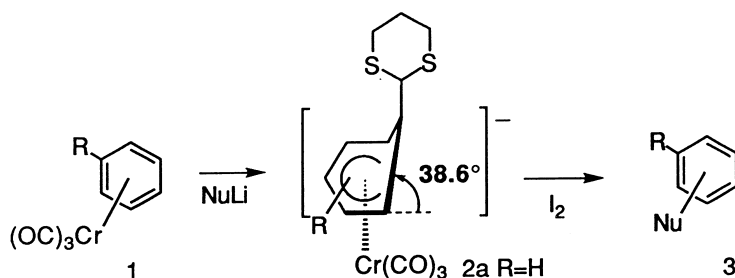
Abstract

Treatment of X-substituted $(\eta^6\text{-arene})$ tricarbonylchromium complexes with a nucleophile and a proton source afforded new arenetricarbonylchromium complexes via chromium hydride intermediates and elimination of HX (*cine* and *tele* nucleophilic aromatic substitutions). Treatment of $(\eta^5\text{-X-substituted-cyclohexadienyl})$ tricarbonyl manganese complexes with hydride and a proton source gave $\eta^5\text{-cyclohexadienyl}$ complexes resulting from a cleavage of the C–X bond: the $\eta^3\text{-X-substituted cyclohexenyl}$ intermediates underwent elimination of an agostic hydrogen and the X group. © 1998 Elsevier Science S.A. All rights reserved.

Keywords: Arene; Arene complexes; Aromatic substitution; Benzenetricarbonylchromium; Benzenetricarbonylmanganese; Nucleophilic substitution

* Corresponding author. Tel: 33 44 27 62 35; Fax: 33 44 27 70 89; e-mail rose@ccrjussieu.fr

¹ Also corresponding author



Scheme 1. Addition of a carbanion to an arenetricarbonylchromium complex

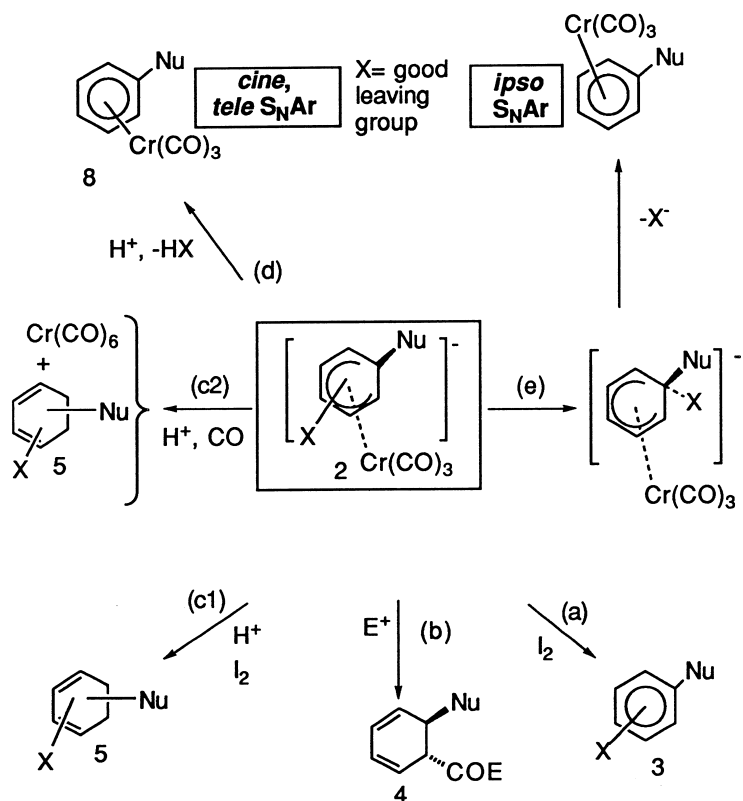
1. Introduction

Tricarbonylchromium and manganese π -arene complexes have found significant applications particularly in organic synthesis [1–36]. They are easily prepared (for Cr-complex preparations see [37–39], for Mn-complex preparations see [40–42]) and the arene group is readily released from complexation in the case of chromium complexes. The arene ligand is a six-electron system occupying three coordination sites and the electron withdrawing power of the Cr(CO)₃ entity is similar to that of the nitro group of nitrobenzene. A decreased electron density of the arene ring is confirmed by the ease of addition of a carbanion [1–36]. This electron density decrease is more significant in the case of cationic tricarbonylmanganese π -complexes. Indeed a large variety of carbanions [43–68] can be added to these manganese complexes, for example ketone enolates and Grignard reagents (see for example [69,70]).

2. Arenetricarbonylchromium complexes

Preparation of arenetricarbonylchromium complexes is easily achieved in good yield usually by heating the free arene with Cr(CO)₆ in a high boiling solvent [37–42]. Addition of a carbanion, first reported by Card and Trahanovsky in the case of a nonhalogenated arene (^tBuLi and ethylbenzenetricarbonylchromium) followed by Ce^{IV} oxidation yielded 2 isomeric ^tbutylethylbenzene [71]. This reaction has been extended in the case of a large variety of nucleophiles. The first step which is the exo addition of the nucleophile to the face opposite to the tricarbonylchromium entity has been definitively proved by the X-ray structure of the anionic intermediate complex. Indeed, adduct of 2-lithio-1,3-dithiane and benzenetricarbonylchromium **1** afforded an anionic complex with an η^5 -cyclohexadienyl structure **2a** (Scheme 1). The deformation of the sp² carbon from the planarity is 38.6° [72]. We will show the key role of this anionic (η^5 -cyclohexadienyl)tricarbonylchromium complex **2**, the fate of which depends on the experimental conditions used. Some of these reactions are summarised in Scheme 2.

Oxidation of **2** releases the substituted arene **3** and the chromium is oxidized [72], whereas reaction of **2** with particular electrophiles yields *trans* disubstituted cyclohex-



Scheme 2. Reactivity of the anionic complex 2

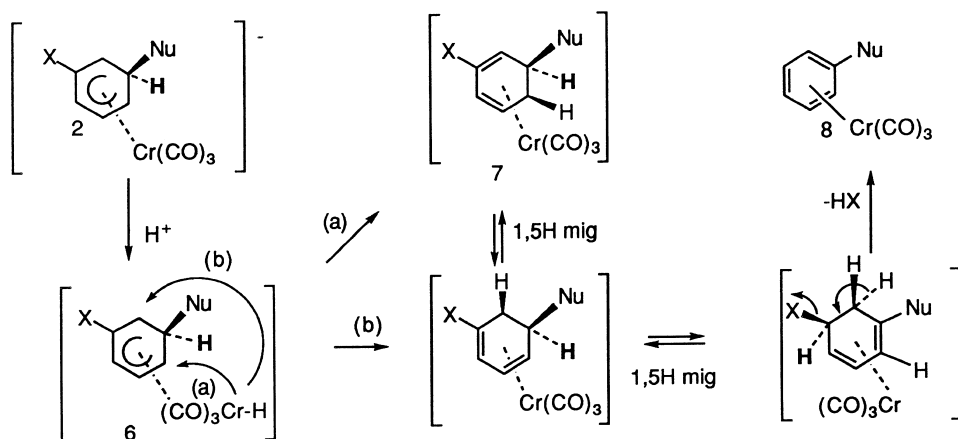
adienes **4** with or without CO insertion (path b, Scheme 2). These reactions have been recently extended into an asymmetric version [73–84].

Treatment of **2** with acid gives a mixture of isomeric substituted cyclohexadienes **5** after iodine oxidation (path c1, Scheme 2) [72]. Treatment of **2** with acid under CO atmosphere yields also isomeric substituted cyclohexadienes **5** (path c2, Scheme 2) but Cr(CO)_6 is recovered and can be used to prepare starting arenetricarbonylchromium complexes [85]. In this case, we have never succeeded in isolating (η^4 -cyclohexadiene)chromium complexes. The formation of compound **5** involves a chromium-hydride after acidic treatment of complex **2** (the presence of an agostic bond is not excluded [79]).

In this review, we focus on the reactivity of anionic complexes such as **2** in acidic medium when they are substituted by good leaving groups (path d, Scheme 2).

2.1. Tele–meta nucleophilic aromatic substitution

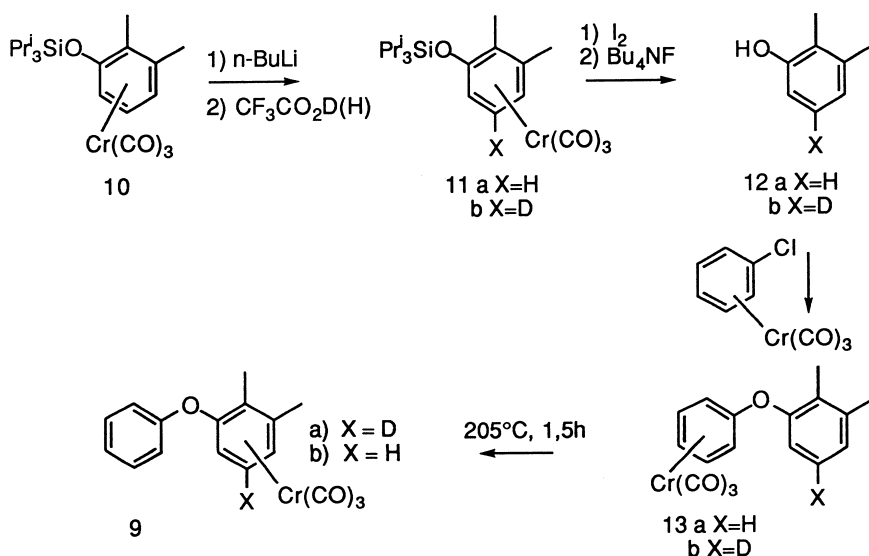
When we started this study, we thought that if complex **2** was substituted by a good leaving group X, we could expect, after a proper treatment, an elimination of HX in order to recover a new arenetricarbonylchromium complex (Scheme 3) [86].



Scheme 3. Tele-meta nucleophilic aromatic substitution

Indeed treatment of complex **2** with a strong acid yields complex **8** whose formation can be explained as follows: protonation of the anionic compound **2** affords the chromium-hydride complex **6** (Scheme 3). Hydride migration to the η^5 -cyclohexadienyl ring gives rise to an η^4 -cyclohexadiene complex **7**. Then isomerization of this cyclohexadiene, thanks to 1,5 H-migrations, allows elimination of HX which rearomatizes the ring: this is the driving force of the reaction.

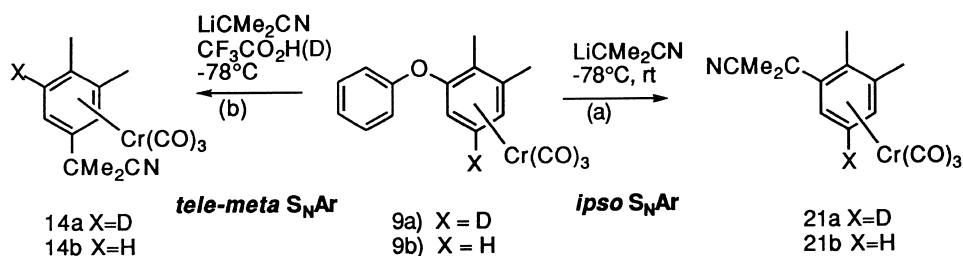
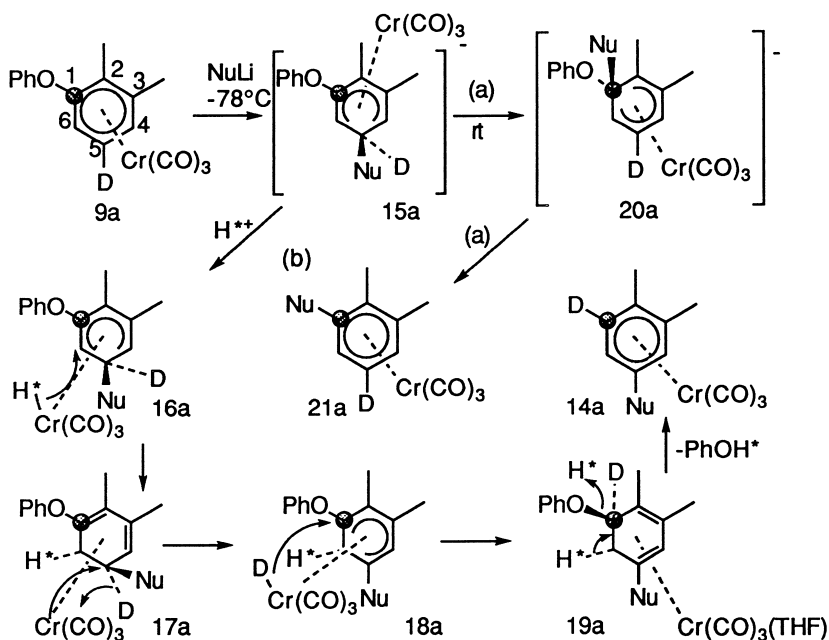
In order to prove this migration, we prepared a labelled diphenylethertricarbo-nylchromium complex **9a** (Scheme 4) substituted by two methyl groups at the C-2 and C-3 positions in order to avoid addition of the nucleophile to these carbons

Scheme 4. Preparation of labelled diphenylethertricarbo-nylchromium complex **9b**

[87]. Treatment of 1-triisopropylsilyloxy-2,3-dimethyl-benzene complex **10**, with *n*-BuLi and CF₃CO₂D gives the 5-deutero complex **11b** (*meta*-lithiation with respect to the silyloxy group). 2,3-Dimethyl-5-deuteriophenol **12b**, obtained by iodine oxidation of complex **11a** followed by Bu₄NF treatment, reacts with chlorobenzene-tricarbonylchromium to yield the diphenylether complex **13b**.

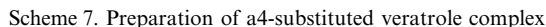
Migration of the $\text{Cr}(\text{CO})_3$ entity from one cycle to the other one is accomplished by heating crystals of complex **12b** at high temperature. The same experiments with $\text{X}=\text{H}$ give complex **9b** (Scheme 4). Complexes **9a** and **9b** treated with Me_2CLiCN and $\text{CF}_3\text{CO}_2\text{H}(\text{D})$ generate complexes **14a** and **14b** (Scheme 5).

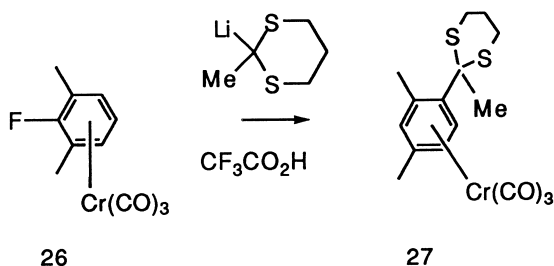
These results can be easily explained. The first step is the addition of the nucleophile Nu[−] to the C5 carbon (Scheme 6). The anionic complex **15a**, trapped with

Scheme 5. *Tele-meta* nucleophilic aromatic substitution

Scheme 6. Tele-meta SnAr: 1,5 deuteride migration

2,6-Dimethylfluorobenzenetricarbonylchromium complex **26** reacts with 2-lithio-2-methyl-1,3-dithiane and $\text{CF}_3\text{CO}_2\text{H}$ and affords the trisubstituted complex **27** via



Scheme 9. *Tele-para* S_NAr

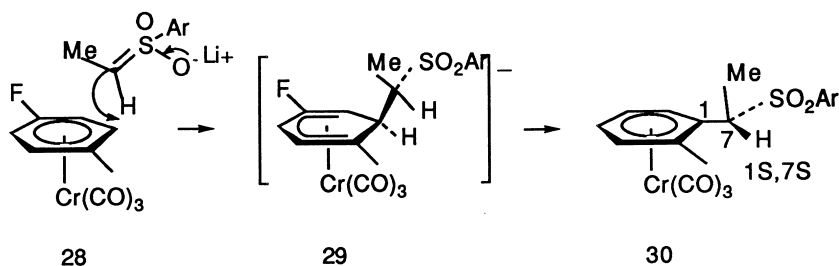
a *tele-meta* S_NAr (Scheme 9) [94]. No formation of 1,3,5 trisubstituted complex is detected.

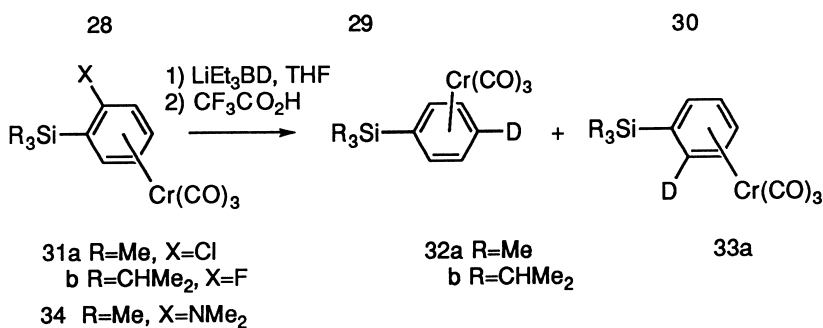
In the case of *para*-fluoro-toluenetricarbonylchromium complex **28**, addition of α -sulfonyl carbanion followed by acid treatment yields only one diastereoisomer **30** via the anionic complex **29**. This reaction sheds light on some aspects of dynamic stereochemistry. We proposed an approach of the carbanion which minimizes the interactions between the carbanion and the starting complex (Scheme 10). This could explain the formation of the SS/RR complex **30** [95,96] (for nomenclature used see [97,98]).

We next considered the question of the reactivity of a very simple nucleophile: an hydride. So, *ortho*-halogeno-trialkylsilylbenzenetricarbonylchromium complex **31a** or **31b** reacts with deuteride (Et_3BDLi) affording, after CF_3CO_2H treatment, deuterio-trimethylsilyl-benzenetricarbonylchromium complexes **32a** and **33a** (90/10) via a *tele-meta* nucleophilic aromatic substitution, or deuterio-triisopropylsilyl-benzenetricarbonylchromium complex **32b** (Scheme 11) [99].

Cleavage of carbon–chloride, carbon–fluoride or carbon–oxygen bonds is well preceded in organic chemistry [100], but cleavage of a carbon–nitrogen bond is much more difficult. Thus we tried reaction of hydrides with aniline complexes. Addition of deuteride ($LiEt_3BD$) to a THF solution of *ortho*-trimethylsilyl-*N,N*-dimethylanilinetricarbonylchromium complex **34** yields deuterio-trimethylsilyl benzenetricarbonylchromium complexes **32a** and **33a** (20/80) after CF_3CO_2H treatment, both resulting from *tele-meta* S_NAr (Scheme 11) [101].

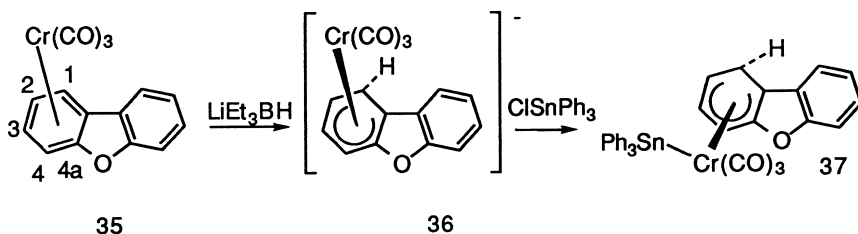
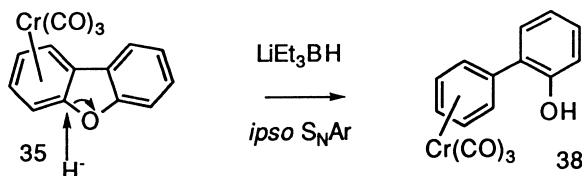
In order to shed light on the mechanism of hydride addition to an arenetricarbo-

Scheme 10. *Tele-meta* S_NAr

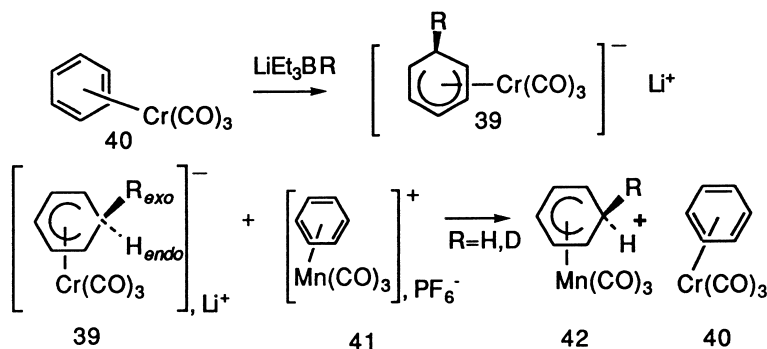
Scheme 11. *Tele-meta* SnAr

nylchromium complex, we undertook different studies with disubstituted complexes. Treatment of dibenzofurantricarboxyl chromium complex **35** with LiEt_3BH yields the anionic compound **36** which can be obtained as a yellow powder [102,103]. Addition of triphenyltinchloride to this anionic intermediate **36** gives the dinuclear complex **37** whose X-ray structure confirms the regioselective addition of hydride to the C-1 carbon at low temperature (Scheme 12). It is worth noting that addition of LiEt_3BH at 67 °C to complex **35** gives, after hydrolysis, the biphenyl derivative **38** (Scheme 13, *ipso* $\text{S}_\text{N}\text{Ar}$). This clearly shows that the *meta*-addition occurs under reversible conditions, complex **36** undergoing an isomerization that ultimately leads to C–O bond *ipso* cleavage and liberation of complex **38**.

We used an other method to establish the reversible addition of hydride to an arenetricarbonylchromium complex: complex **39**, obtained quantitatively by adding LiEt_3BH to benzenetricarbonylchromium complex **40** reacted with cationic benzenetricarbonylmanganese complex **41** [104] to give a mixture of complexes **40** and

Scheme 12. Trapping of the anionic complex by ClSnPh_3 

Scheme 13. Addition of hydride to dibenzofuran complex

Scheme 14. Reversible addition of hydride to **40**

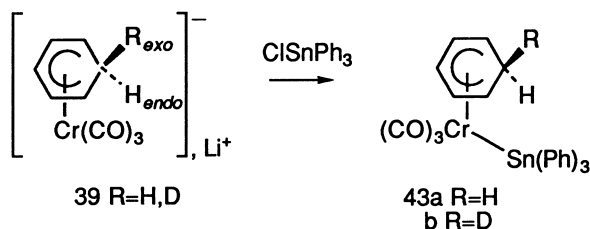
neutral η^5 -cyclohexadienylmanganesetricarbonyl **42**. This is the result of an hydride transfer from **39** toward the electrophilic complex **41** (Scheme 14) [102,103].

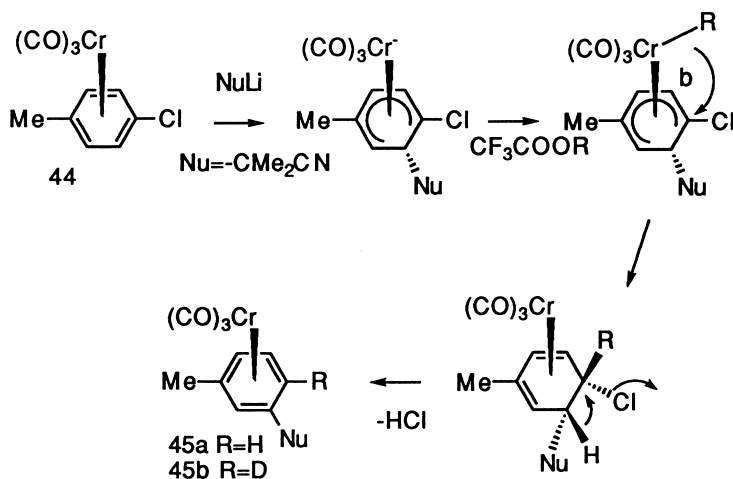
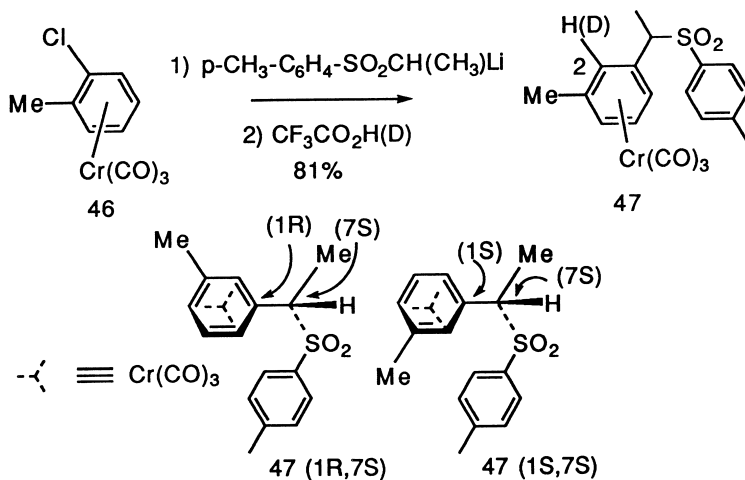
Anionic complexes **39** whose ^1H NMR spectrum can be achieved in THF-*d*8 in a sealed tube, (R=H, D) can also be trapped by triphenyltin chloride giving the Sn–Cr complex **43a** (R=H) (Scheme 15) or **43b** [102,103].

2.2. *Cine* nucleophilic aromatic substitution

We next considered the question of another $\text{S}_{\text{N}}\text{Ar}$ in which the entering group Nu might take up a position *ortho* to the leaving group (a chloride atom or a methoxy group for example) [105,93] (a few examples of *cine* substitutions have been recently described [106]). We studied the addition of 2-lithioisobutyronitrile to **44** followed by $\text{CF}_3\text{CO}_2\text{H}$ treatment. Complex **45a** is obtained as the major product (Scheme 16). This reaction clearly shows that the carbanion addition occurs *ortho* to the leaving group and *meta* to the methyl group. When $\text{CF}_3\text{CO}_2\text{D}$ is used, the deuterio complex **45b** is obtained in a good agreement with a mechanism, involving hydride migration, then HCl elimination as depicted in Scheme 16.

In the case of complex **46**, addition of the anion of ethyl-*para*-tolylsulfone affords, after acid treatment, one regioisomer constituted by only one diastereoisomer **47** (Scheme 17). If $\text{CH}_3\text{CO}_2\text{D}$ is used, complex **47** is deuterated at the C-2 carbon in a good agreement with a *cine* $\text{S}_{\text{N}}\text{Ar}$. But we were not able to obtain an X-ray structure

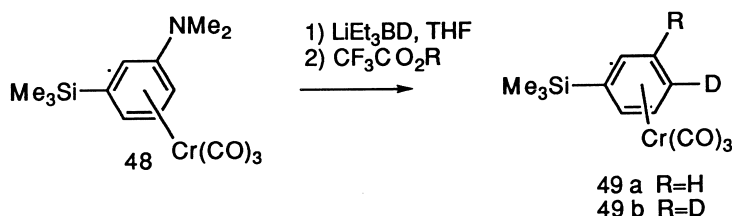
Scheme 15. Formation of a dinuclear Cr–Sn(η^5 -cyclohexadienyl) complex

Scheme 16. *Cine* $\text{S}_{\text{N}}\text{Ar}$ Scheme 17. *Cine* $\text{S}_{\text{N}}\text{Ar}$

in order to know which one of the two diastereoisomers: 1*R*, 7*S*, (1*S*, 7*R*) or 1*S*, 7*S* (1*R*, 7*R*) was obtained (Scheme 17) [95–98]. In the case of *meta*-trimethylsilyl, *N,N*-dimethylanilinetricarbonylchromium complex **48**, complex **49a** (or **49b**) is obtained under the same experimental conditions (Scheme 18) [101].

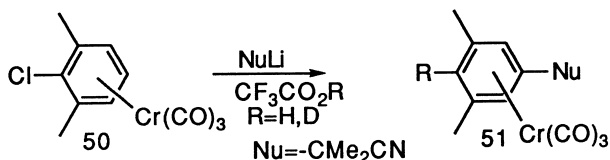
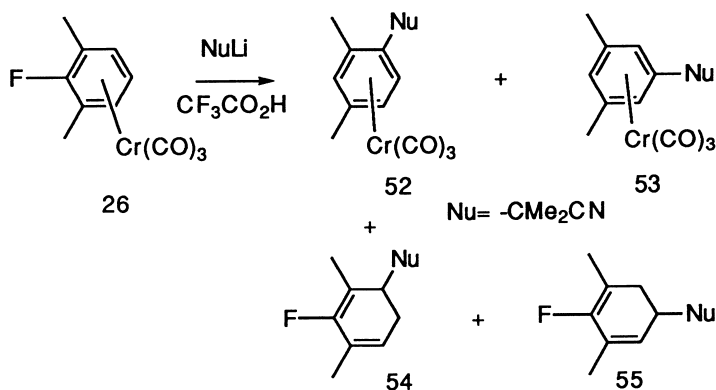
2.3. *Tele*–*para* nucleophilic aromatic substitution

A third new substitution, namely a *tele*–*para* $\text{S}_{\text{N}}\text{Ar}$, was discovered in the case of 2,6-dimethylchlorobenzenetricarbonylchromium complex **50**. Reaction of 2-lithio,

Scheme 18. *Cine* SnAr

2-methylpropionitrile with complex **50** affords after quenching with $\text{CF}_3\text{CO}_2\text{H(D)}$ a single arenetricarbonylchromium complex **51** via a *tele-para* nucleophilic aromatic substitution (Scheme 19) [107].

Reaction of 2-lithio-2-methylpropionitrile with 2,6-dimethylfluorobenzene **26** gives rise to the formation of complexes **52**, **53** and cyclohexadienes **54**, **55**. The same reaction is more regioselective in the presence of HMPA, indeed it yields cyclohexadienes **54** and **55** (1/3) and only complex **53** (Scheme 20) meaning probably that **53** and **55** are formed under kinetic control [108–111]. Thus, thanks to these last examples, we concluded that the entering groups could take up a position two atoms away from the atom to which the leaving group was attached.

Scheme 19. *Para-tele* SnArScheme 20. *Tele-meta* and *tele-para* SnAr

3. Cationic arenetricarbonylmanganese complexes

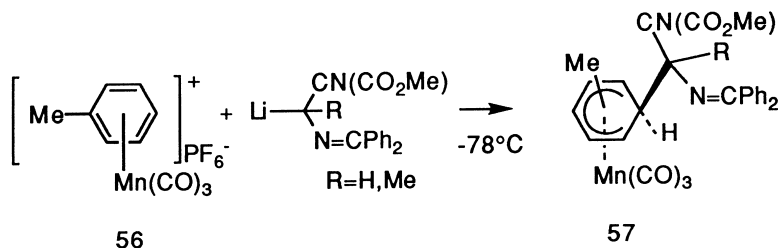
The activation of arenes by coordination to transition metals provides a synthetic route to functionalized arenes in the case of Cr. The manganese mediated functionalization of arenes is also attractive because of the high electrophilicity of the corresponding compounds. A wide variety of nucleophiles can be added to the arene ring [43– 68]. We decided to study not only the reactivity of cationic (η^6 -arene)tricarbonylmanganese complexes, but also of neutral substituted (η^5 -cyclohexadienyl) tricarbonyl manganese complexes.

3.1. Reactivity of cationic arenetricarbonylmanganese complexes.

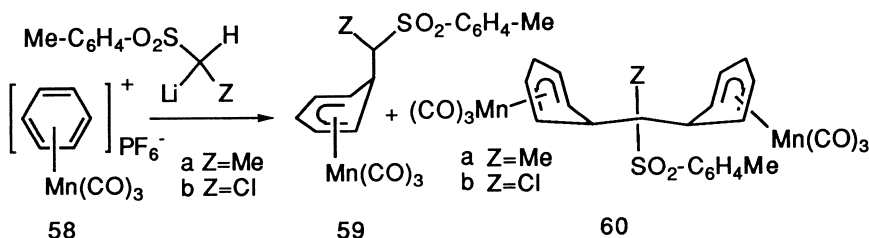
Nucleophilic aromatic addition of stabilized carbanions give almost quantitatively corresponding stable η^5 -neutral cyclohexadienyl manganese complexes. α -Iminoesters or iminonitriles carbanions react with toluenetricarbonylmanganese **56** and give the η^5 -cyclohexadienyl complexes **57** (Scheme 21). Two regioisomers are obtained (*meta:ortho* = 70:30) [112,113].

α -Sulfonyl or α -chloro-sulfonyl carbanions react with the cationic benzene complex **58** to yield as the major product an unexpected bimetallic compound **60** whose formation can be explained by the addition of the benzylic carbanion obtained from the neutral complex **59** with the starting compound **58** (Scheme 22) [112,113]. α -Nitro carbanions react with *para*-chlorotoluenetricarbonylmanganese **61** and give neutral complexes **62** and **63** (Scheme 23).

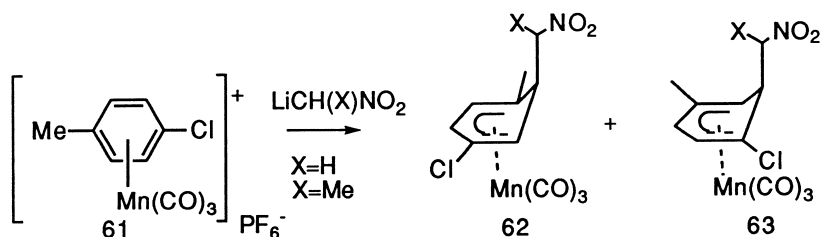
α -Chloroester, α -bromoester, α -isocyanoester, nitrile carbanions react nearly quan-



Scheme 21. Addition of imino-nitrile and ester carbanions



Scheme 22. Addition of α -sulfonyl carbanions

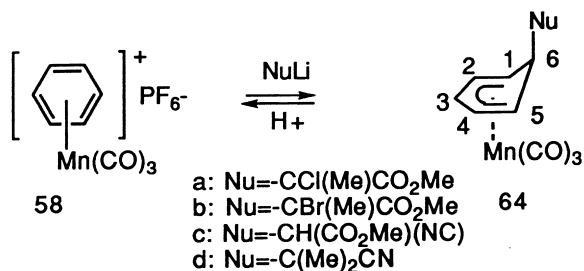


Scheme 23. Addition of x-nitro carbanions

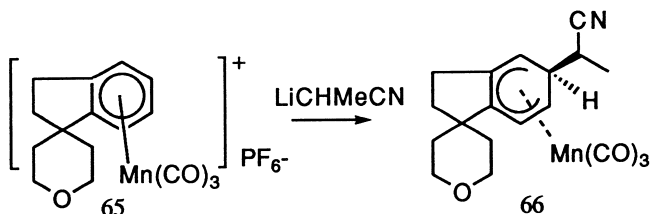
titatively with the hexafluorophosphate of benzenetricarbonylmanganese **58** to give the corresponding neutral η^5 -cyclohexadienylmanganese complex **64** (Scheme 24) [114,115]. The X-ray structure of complex **64a** showed an (η^5 -cyclohexadienyl) structure. The cyclohexadienyl ring is nearly planar and folded about $\text{C}_1\text{C}_6\text{C}_5$ with an angle of 35° . The sp^3 C_6 carbon of the ring is eclipsed by a Mn-CO bond in agreement with other (η^5 -cyclohexadienyl) complex structures [115].

More sophisticated cationic (η^6 -arene) complex such as spiroindane derivative **65** has been synthesized and was reacted with 2-lithio-2-methylpropionitrile. We obtained, as the major product, the regioisomer **66** (Scheme 25) [114]. This regio-selective reaction could be of interest if applied to the synthesis of 2-aryl propionic acids which are known to have anti-inflammatory properties.

Knowing that ketone enolates and anionic Fischer type carbenes have similar reactivity, we added Fischer type carbene carbanions **67** to the cationic arene complex



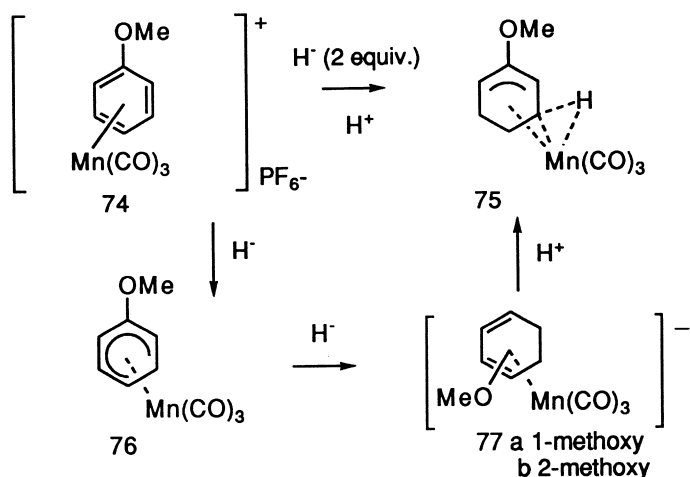
Scheme 24. Addition of carbanions

Scheme 25. Addition of propionitrile carbanion to **65**

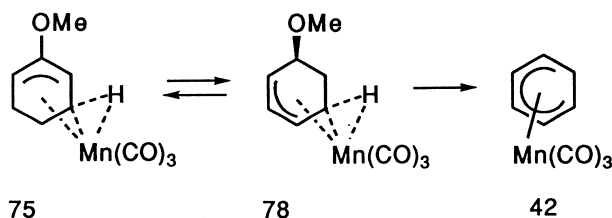
3.2. Reactivity of neutral (η^5 -cyclohexadienyl) tricarbonylmanganese complexes

Facile isomerization of (η^3 -cyclohexenyl)tricarbonylmanganese complexes involving Mn–H–C agostic bond (two electron, three-center Mn–H–C bond) has been described by Brookhart et al. [118]. These compounds can be prepared from η^6 or η^5 complexes. Thus complex **75** is obtained by treating the cationic anisoletricarbonylmanganese complex **74** with hydride and then with acid (Scheme 29). The first equivalent of hydride affords the neutral η^5 -2-methoxy-cyclohexadienyl complex **76**. Addition of a second equivalent of hydride ($\text{KB}(\text{O}-i\text{-Pr})_3\text{H}$) to this complex generates 1-methoxy and 2-methoxycyclohexa-1,3-diene anions **77**. Using L-selectride as hydride source, we obtained only the 2-methoxy isomer **77b** [119]. Addition of acid gives as the sole isomer the 2-methoxy-(η^3 -cyclohexenyl)manganesetricarbonyl complex **75** via protonation of the anionic complex **77** followed by migration of the hydrogen to the proximal face of the ring.

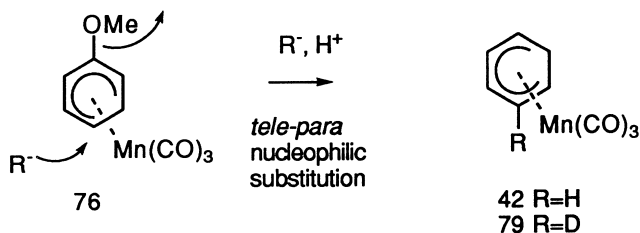
We wanted to obtain the isomerization of complex **75** in such a way that MeOH elimination would be possible to give rise to the formation of an other η^5 -cyclohexadienyl complex. We showed that isomerization of complex **75** could occur after heating it in THF-*d*8 at 70 °C in a sealed NMR tube. In the absence of acid or base, no MeOH elimination is observed. In the presence of MeOH or $\text{CF}_3\text{CO}_2\text{H}$ for example, heating complex **75** in THF for 3 h results in the formation of (η^5 -cyclohexadienyl)tricarbonylmanganese complex **42** via a new complex **78**. This postulated (η^3 -methoxycyclohexenyl)tricarbonylmanganese intermediate **78** could undergo elimination of an agostic hydrogen and the methoxy group with formation of the η^5 compound **42** (Scheme 30). In other words, complex **42** could be synthesized from compound **76** via substitution of the methoxy group by an hydride (Scheme 31). Because of the position of the entering group (H^-) relative to that of the leaving group (MeO^-), we called this reaction a *tele-para* nucleophilic



Scheme 29. *Tele-para* nucleophilic substitution



Scheme 30. Elimination of methanol

Scheme 31. Addition of hydride and acid to an (η^5 -cyclohexadienyl)manganese complex

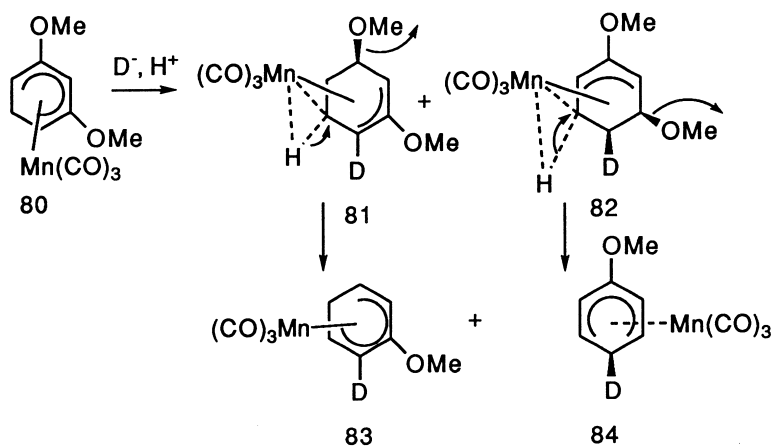
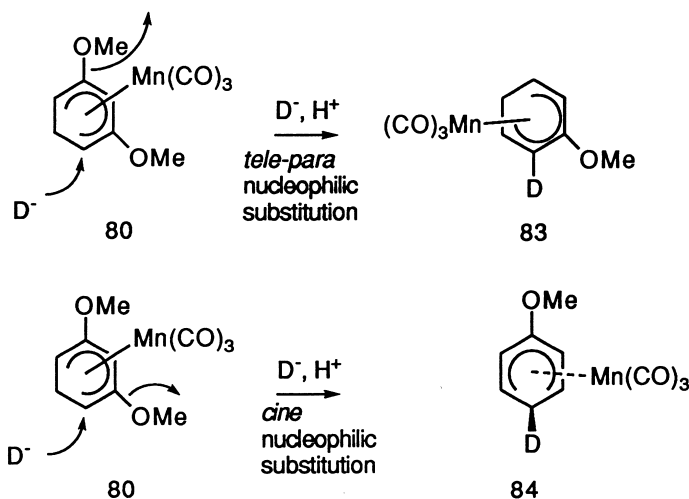
substitution by analogy with the S_NAr in (η^6 -arene)chromium complex series. Using deuteride and (or) labeled acid, we studied the mechanism of this reaction: for example, addition of deuteride to complex **76** gives exclusively deuterated complex **79** after acid treatment and heating (Scheme 31).

We extended this reaction to other (η^5 -X-substituted-cyclohexadienyl) tricarbonyl-manganese complexes (X = OR, Cl, NMe₂, SPh) which gave, under the same conditions, (η^5 -cyclohexadienyl) complexes resulting from a cleavage of the C–O, C–Cl, C–N and C–S bonds after hydride and a proton source treatment.

For example complex **80** affords complexes **81** and **82** after deuteride and acid treatment (Scheme 32). By heating these η^3 -cyclohexadienyl compounds, elimination of MeOH occurs and gives rise to the formation of the labeled complexes **83** and **84**, the formation of which can be explained by an overall *tele-para* substitution for complex **83** and *cine* for complex **84** (Schemes 32 and 33) [119].

These reactions represent the first *cine* and *tele*-substitutions in the case of Mn complexes.

In summary, arenetricarbonylchromium and manganese complexes allow the addition of a wide variety of nucleophiles under mild conditions. In the light of our results, it is clear that the three new nucleophilic substitutions (*cine*, *tele-meta* and *tele-para*) which have been discovered by our group, broaden the scope of the applications of these complexes in organic chemistry. The intermediates involve chromium-hydrogen and manganese-hydrogen (or agostic) bonds. The organometallic entity Cr(CO)₃ or Mn(CO)₃ remains coordinated either to the arene cycle or to the cyclohexadienyl ring.

Scheme 32. *Tele-para* and *cine* nucleophilic substitutionsScheme 33. *Tele-para* and *cine* nucleophilic substitutions

Acknowledgements

For the published and unpublished work from our group, grateful acknowledgment is given to an enthusiastic group of collaborators: K. Aniss, F. Balssa, O. Bellot, J.C. Boutonnet, R. Chavignon, J.P. Djukic, J. Garcia-Oricain, P. Geyserians, R. Khourzom, F. Landres, C. Le Corre-Susanne, L. Mignon, L. Mordenti, J.M. Normant, V. Onnikian, A. Perrotey, L. Piumi, A. Semra, F. Simon, J.P. Souchez, R. Valentic, F. Teldji, J.P. Tranchier.

References

- [1] G. Jaouen, *Ann. N. Y. Acad. Sci.* 295 (1977) 59.
- [2] M.F. Semmelhack, *Ann. N. Y. Acad. Sci.* 295 (1977) 36.
- [3] M.F. Semmelhack, G.R. Clark, J.L. Garcia, J.J. Harrison, Y. Thebtaranonth, W. Wulff, A. Yamashita, *Tetrahedron* 37 (1981) 3957.
- [4] S.G. Davies, *Organotransition Metal Chemistry, Applications to Organic Syntheses*, Pergamon Press, Oxford, 1982, p. 166.
- [5] W.E. Watts, *Comprehensive Organometallic Chemistry*, vol. 8, Pergamon Press, Oxford, 1982, p. 1013.
- [6] A. Solladie-Cavallo, *Polyhedron* 4 (1985) 901.
- [7] V.N. Kalinin, *Russian Chem. Rev.* 56 (1987) 682.
- [8] J.P. Collman, L.S. Hegedus, J.R. Norton, R.G. Finke, *Principles and Applications of Organotransition Metal Chemistry*, vol. 20, University Science Books, MW Valley, CA, 1987, p. 921.
- [9] L. Balas, D. Jhurry, L. Latxague, S. Grelier, Y. Morel, M. Hamdani, D. Astruc, *Bull. Soc. Chim. Fr.* 127 (1990) 401.
- [10] M.F. Semmelhack, *Comprehensive Organic Synthesis*, vol. 4, B.M. Trost (Ed.), Pergamon Press, Oxford, 1991, 517.
- [11] F.J. Mc Quillin, D.G.N. Parker, G.R. Stephenson, *Transition Metal in Organic Synthesis*, Cambridge University Press, Cambridge, 1991.
- [12] M.C. Sénéchal-Tocquer, D. Sénéchal, J.Y. Le Bihan, D. Gentric, B. Caro, *Bull. Soc. Chim. Fr.* 129 (1992) 121.
- [13] F. Rose-Munch, E. Rose, *Trends in Organometallic Chemistry, Research Trends*, vol. 1, Trivandrum, India, 1994, 669.
- [14] M.F. Semmelhack, in: E.W. Abel, F.G.A. Stone, G. Wilkinson (Eds.), *Comprehensive Organometallic Chemistry II*, vol. 12, Pergamon Press, Oxford, 1995, p. 979.
- [15] S.G. Davies, T.D. McCarthy, in: E.W. Abel, F.G.A. Stone and G. Wilkinson (Eds.), *Comprehensive Organometallic Chemistry II*, vol. 12, Elsevier, Oxford, 1995, p. 1039.
- [16] M.F. Semmelhack, G. Clark, *J. Am. Chem. Soc.* 99 (1977) 1675.
- [17] M.F. Semmelhack, J. Bisaha, M. Czarny, *J. Am. Chem. Soc.* 101 (1979) 768.
- [18] M.F. Semmelhack, W. Wulff, J.L. Garcia, *J. Organomet. Chem.* 240 (1982) C5.
- [19] M. Uemura, N. Nishikawa, K. Take, M. Ohnishi, K. Hirotsu, T. Higuchi, Y. Hayashi, *J. Org. Chem.* 48 (1983) 2349.
- [20] N.F. Masters, A. Widdowson, *J. Chem. Soc., Chem. Comm.* (1983) 955.
- [21] P.J. Beswick, S.J. Leach, N.F. Masters, D.A. Widdowson, *J. Chem. Soc., Chem. Comm.* (1984) 46.
- [22] P.L. Beswick, D.A. Widdowson, *Synthesis* (1985) 492.
- [23] M.F. Semmelhack, H.G. Schmalz, *Tetrahedron Lett.* 37 (1996) 3089.
- [24] A.J. Pearson, A.V. Gontcharov, P.D. Woodgate, *Tetrahedron Lett.* 37 (1996) 3087.
- [25] J.C. Boutonnet, L. Mordenti, E. Rose, O. Le Martret, G. Precigoux, *J. Organomet. Chem.* 221 (1981) 147.
- [26] J.C. Boutonnet, E. Rose, *J. Organomet. Chem.* 221 (1981) 157.
- [27] C. Guimon, G. Pfister-Guillouzo, E. Rose, *J. Organomet. Chem.* 224 (1982) 125.
- [28] J.C. Boutonnet, J. Levisalles, E. Rose, G. Precigoux, C. Courseille, N. Platzter, *J. Organomet. Chem.* 255 (1983) 317.
- [29] J.C. Boutonnet, J. Levisalles, F. Rose-Munch, E. Rose, G. Precigoux, F. Leroy, *J. Organomet. Chem.* 290 (1985) 153.
- [30] J.C. Boutonnet, F. Rose-Munch, E. Rose, Y. Jeannin, F. Robert, *J. Organomet. Chem.* 297 (1985) 185.
- [31] J.C. Boutonnet, F. Rose-Munch, E. Rose, G. Precigoux, *J. Organomet. Chem.* 284 (1985) C25.
- [32] J. Levisalles, F. Rose-Munch, E. Rose, A. Semra, J. Garcia-Oricain, Y. Jeannin, F. Robert, *J. Organomet. Chem.* 328 (1987) 109.
- [33] F. Rose-Munch, K. Aniss, E. Rose, *J. Organomet. Chem.* 385 (1990) C1.
- [34] F. Rose-Munch, K. Aniss, E. Rose, J. Vaisserman, *J. Organomet. Chem.* 415 (1991) 223.

- [35] F. Rose-Munch, L. Mignon, J.P. Souchez, *Tetrahedron Lett.* 32 (1991) 6323.
- [36] L. Besson, M. Le Bail, D.J. Aitken, H.P. Husson, F. Rose-Munch, E. Rose, *Tetrahedron Lett.* 37 (1996) 3307.
- [37] W. Strohmeier, *Chem. Ber.* 94 (1961) 2490.
- [38] C.A. L. Mahaffy, P.L. Pauson, *Inorg. Synth.* 19 (1979) 154.
- [39] M. Hudecek, V. Gadjia, S. Toma, *J. Organomet. Chem.* 413 (1991) 155–161.
- [40] A.J. Pearson, J. Richards, *J. Organomet. Chem.* 258 (1983) C41.
- [41] M.I. Rybinskaya, V.S. Kaganovich, *Russian Chem. Bull.* 42 (1993) 1128.
- [42] S. Sun, L.K. Yeung, D.A. Sweigart, T.Y. Lee, S.S. Lee, Y.K. Chung, S.R. Switzer, R.D. Pike, *Organometallics* 14 (1995) 2613.
- [43] P.J.C. Walker, R.J. Mawby, *J. Chem. Soc., Chem. Comm.* (1972) 330.
- [44] P.J.C. Walker, R.J. Mawby, *Inorg. Chim. Acta* 7 (1973) 621.
- [45] P.L. Pauson, J.A. Segal, *J. Chem. Soc., Dalton Trans.* (1975) 1677.
- [46] Y.K. Chung, P.G. Williard, D.A. Sweigart, *Organometallics* 1 (1982) 1053.
- [47] A.J. Pearson, I.C. Richards, *J. Organomet. Chem.* 258 (1983) C41.
- [48] L.A.P. Kane-Maguire, E.D. Honig, D.A. Sweigart, *Chem. Rev.* 84 (1984) 525.
- [49] Y.K. Chung, D.A. Sweigart, N.G. Connelly, J.B. Sheridan, *J. Am. Chem. Soc.* 107 (1985) 2388.
- [50] W.H. Miles, P.M. Smiley, H.R. Brinkman, *J. Chem. Soc., Chem. Comm.* (1989) 1897.
- [51] Y.A. Lee, Y.K. Chung, Y. Kim, J.H. Jeong, *Organometallics* 9 (1990) 2851.
- [52] R.D. Pike, D.A. Sweigart, *Synlett.* (1990) 565.
- [53] A.J. Pearson, S.H. Lee, F. Gouzoules, *J. Chem. Soc., Perkin Trans.* (1990) 2251.
- [54] Y.-A. Lee, Y.K. Chung, Y. Kim, J.H. Jeong, G. Chung, D. Lee, *Organometallics* 10 (1991) 3707.
- [55] W.H. Miles, H.R. Brinkman, *Tetrahedron Lett.* 33 (1992) 589.
- [56] E. Jeong, Y.K. Chung, *J. Organomet. Chem.* 434 (1992) 225.
- [57] A.S. Oh, Y.K. Chung, S. Kim, *Organometallics* 11 (1992) 1394.
- [58] T.Y. Lee, Y.K. Kang, R.D. Pike, D.A. Sweigart, *Inorg. Chim. Acta* 214 (1993) 125.
- [59] W.J. Ryan, P.E. Peterson, Y. Cao, P.G. Williard, D.A. Sweigart, C.D. Baer, C.F. Thompson, Y.K. Chung, T.M. Chung, *Inorg. Chim. Acta* 211 (1993) 1.
- [60] S.G. Lee, Y.K. Chung, T.S. Yoon, W. Shin, *Organometallics* 12 (1993) 2873.
- [61] R. Réau, R.W. Reed, F. Dahan, G. Bertrand, *Organometallics* 12 (1993) 1501.
- [62] K. Woo, G.B. Carpenter, D.A. Sweigart, *Inorg. Chim. Acta* 220 (1994) 297.
- [63] K. Woo, P.G. Williard, D.A. Sweigart, N.W. Duffy, B.H. Robinson, J. Simpson, *J. Organomet. Chem.* 487 (1995) 111.
- [64] T.Y. Lee, S.S. Lee, Y.K. Chung, S.W. Lee, *J. Organomet. Chem.* 486 (1995) 141.
- [65] D.K. Astley, S.T. Astley, *J. Organomet. Chem.* 487 (1995) 253.
- [66] Y. Cao, K. Woo, L.K. Yeung, G.B. Carpenter, D.A. Sweigart, *Organometallics* 16 (1997) 178.
- [67] S. Sun, C.A. Dullaghan, D.A. Sweigart, *J. Chem. Soc., Dalton Trans.* (1996) 4493.
- [68] C.A. Dullaghan, G.B. Carpenter, D.A. Sweigart, *Chem. Eur. J.* 3 (1997) 75.
- [69] Y.K. Chung, P.G. Williard, A. Dwight, D.A. Sweigart, *Organometallics* 1 (1982) 1053.
- [70] R.P. Alexander, G. R Stephenson, *J. Organomet. Chem.* 314 (1986) C73.
- [71] R.J. Card, W.S. Trahanovsky, *Tetrahedron Lett.* 39 (1973) 3823.
- [72] M.F. Semmelhack, Jr., H.T. Hall, R. Farina, M. Yoshifuji, G. Clark, T. Bargar, K. Hirotsu, J. Clardy, *J. Am. Chem. Soc.* 101 (1979) 3535.
- [73] E.P. Kündig, *Pure Appl. Chem.* 57 (1985) 1855.
- [74] E.P. Kündig, A.F., Jr, Cunningham P. Paglia, D.P. Simmons, *Helv. Chim. Acta* 73 (1990) 386.
- [75] E.P. Kündig, M. Inage, G. Bernardinelli, *Organometallics* 10 (1991) 2921.
- [76] E.P. Kündig, G. Bernardinelli, R. Liu, A. Ripa, *J. Am. Chem. Soc.* 113 (1991) 9676.
- [77] E.P. Kündig, A. Ripa, G. Bernardinelli, *Ang. Chem.* 31 (1992) 1071.
- [78] E.P. Kündig, R. Liu, A. Ripa, *Helv. Chim. Acta* 75 (1992) 2657.
- [79] E.P. Kündig, D. Amurrio, G. Bernardinelli, R. Chowdhury, *Organometallics* 12 (1993) 4275.
- [80] E.P. Kündig, A. Ripa, R. Liu, G. Bernardinelli, *J. Org. Chem.* 59 (1994) 4773.
- [81] D. Amurrio, K. Khan, E.P. Kündig, *J. Org. Chem.* 61 (1996) 2258.
- [82] E.P. Kündig, A. Quattropiani, M. Inage, A. Ripa, C. Dupré, A.F., Jr, Cunningham B. Bourdin, *Pure Appl. Chem.* 68 (1996) 97.

- [83] D. Beruben, E.P. Kündig, *Helv. Chim. Acta* 79 (1996) 1533.
- [84] A. Quatropiani, G. Anderson, G. Bernardinelli, E.P. Kündig, *J. Amer. Chem. Soc.* 119 (1997) 4773.
- [85] J.C. Boutonnet, J. Levisalles, J.M. Normant, E. Rose, *J. Organomet. Chem.* 255 (1983) C21.
- [86] J.C. Boutonnet, F. Rose-Munch, E. Rose, *Tetrahedron Lett.* 26 (1985) 3989.
- [87] F. Rose-Munch, E. Rose, A. Semra, *J. Chem. Soc., Chem. Comm.* (1986) 1108.
- [88] V. Gold (Ed.), *Glossary of terms used in physical organic chemistry*, *Pure Appl. Chem.* 51 (1979) 1725.
- [89] V. Gagliardini, V. Onnikian, F. Rose-Munch, E. Rose, *Inorganica Chim. Acta* 259 (1997) 265.
- [90] F. Rose-Munch, E. Rose, A. Semra, *J. Organomet. Chem.* (1989) C9.
- [91] M. Persson, V. Hacksell, I. Csoregh, *J. Chem. Soc., Perkin Trans.* (1991) 1453.
- [92] H.G. Schmalz, K. Schellhaas, *Ang. Chem.* 35 (1996) 2146 and references therein
- [93] F. Rose-Munch, O. Bèfrot, L. Mignon, A. Semra, F. Robert, Y. Jeannin, *J. Organomet. Chem.* 402 (1991) 1.
- [94] F. Rose-Munch, E. Rose, A. Semra, M. Mignon, C. Knobler, *J. Organomet. Chem.* 363 (1989) 297.
- [95] R. Khourzom, F. Rose-Munch, E. Rose, *Tetrahedron Lett.* 31 (1990) 2011.
- [96] F. Rose-Munch, R. Khourzom, J.P. Djukic, E. Rose, J. Brocard, *J. Organomet. Chem.* 46 (1994) 195.
- [97] J. Besançon, J. Tirouflet, *Bull. Soc. Chim. Fr.* (1969) 861.
- [98] A. Solladie-Cavallo, G. Solladie, E. Tsamo, *J. Org. Chem.* 44 (1979) 4189.
- [99] J.P. Djukic, P. Geysermans, F. Rose-Munch, E. Rose, *Tetrahedron Lett.* 32 (1991) 6703.
- [100] A. Maercker, *Ang. Chem. Int. Ed. Engl.* 26 (1987) 972.
- [101] J.P. Djukic, F. Rose-Munch, E. Rose, *J. Chem. Soc. Chem. Comm.* 22 (1991) 1634.
- [102] J.P. Djukic, F. Rose-Munch, E. Rose, Y. Dromzee, *J. Am. Chem. Soc.* 115 (1993) 6434.
- [103] J.P. Djukic, F. Rose-Munch, E. Rose, *Organometallics* 14 (1995) 2027.
- [104] G. Winkhaus, L. Pratt, G. Wilkinson, *J. Chem. Soc.* (1961) 3807.
- [105] F. Rose-Munch, E. Rose, A. Semra, *J. Chem. Soc. Chem. Comm.* (1986) 1551.
- [106] H.G. Schmalz, S. Siegel, A. Schwarz, *Tetrahedron Lett.* 37 (1996) 2947 and references therein
- [107] F. Rose-Munch, E. Rose, A. Semra, M. Philoche, *J. Organomet. Chem.* 363 (1989) 123.
- [108] E.P. Kündig, V. Desobry, D.P. Simmons, E. Wenger, *J. Am. Chem. Soc.* 111 (1989) 1804.
- [109] B. Ohlsson, C. Ullén, *J. Organomet. Chem.* 267 (1984) C34.
- [110] B. Ohlsson, C. Ullén, *J. Organomet. Chem.* 350 (1988) 35.
- [111] F. Rose-Munch, E. Rose, A. Semra, L. Mignon, J. Garcia-Oricain, C. Knobler, *J. Organomet. Chem.* 363 (1989) 297.
- [112] F. Rose-Munch, K. Aniss, *Tetrahedron Lett.* 31 (1990) 6351.
- [113] F. Rose-Munch, C. Susanne, C. Renard, E. Rose, J. Vaisserrnan, *J. Organomet. Chem.* 519 (1996) 253.
- [114] F. Balssa, K. Aniss, F. Rose-Munch, *Tetrahedron Lett.* 33 (1992) 1901.
- [115] F. Balssa, V. Gagliardini, C. Le Corre-Susanne, F. Rose-Munch, E. Rose, J. Vaisserrman, *Bull. Soc. Chim. Fr.* 134 (1997) 537.
- [116] F. Rose-Munch, C. Le Corre-Susanne, F. Balssa, E. Rose, J. Vaissennan, E. Licandro, A. Papagni, S. Maiorana, W. D. Meng and G. R. Stephenson, *J. Organomet. Chem.* 1997 (in press).
- [117] V. Gagliardini, F. Balssa, F. Rose-Munch, E. Rose, C. Susanne, Y. Dromzee, *J. Organomet. Chem.* 519 (1996) 281.
- [118] M. Brookhart, M. Green, L. Wong, *Prog. Inorg. Chem.* 36 (1988) 1.
- [119] F. Balssa, V. Gagliardini, F. Rose-Munch, E. Rose, *Organometallics* 15 (1996) 4373.